organic papers

Acta Crystallographica Section E Structure Reports Online

ISSN 1600-5368

John Nicolson Low,^a* Justo Cobo,^b Manuel Nogueras,^b Adolfo Sánchez,^b Braulio Insuasty^c and Yupanqui Caldas^c

^aDepartment of Chemistry, University of Aberdeen, Meston Walk, Old Aberdeen AB24 3UE, Scotland, ^bDepartamento de Química Inorgánica y Orgánica, Universidad de Jaén, 23071 Jaén, Spain, and ^cGrupo de Investigación de Compuestos Heterocíclicos, Departamento de Química, Universidad del Valle, AA25360 Cali, Colombia

Correspondence e-mail: jnlow111@hotmail.com

Key indicators

Single-crystal X-ray study T = 120 KMean σ (C–C) = 0.003 Å R factor = 0.037 wR factor = 0.088 Data-to-parameter ratio = 17.0

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

 \odot 2003 International Union of Crystallography Printed in Great Britain – all rights reserved

Ladders in the supramolecular structure of 1-(4-bromophenyl)-3-[2-(2-hydroxyethyl)amino-5-nitroanilino]-1-propanone dimethylformamide solvate

The title compound, $C_{17}H_{18}BrN_3O_4 \cdot C_3H_7NO$, has a supramolecular structure which is based on two antiparallel C(11)chains, formed by $O-H \cdot \cdot \cdot O$ hydrogen bonds. It is enhanced by a further linkage via two $N-H \cdot \cdot \cdot O$ hydrogen bonds to the DMF solvent molecule, forming an $R_2^1(7)$. $C-H \cdot \cdot \cdot O$ weak hydrogen bonds link adjacent molecules in the chain. These chains are then linked together by two $C-H \cdot \cdot \cdot O$ weak hydrogen bonds forming a molecular ladder, which runs parallel to the *b* axis and comprises a series of ring structures.

Comment

In our ongoing research into the synthesis of new biologically interesting diazepine fused derivatives we attempted the reaction of 3-dimethylamino-1-(4-bromophenyl)-1-propanone, a precursor of the corresponding α , β -unsaturated ketone, with 2-(2-amino-4-nitroanilino)ethanol, a highly functionalized *ortho*-diamino compound, but instead of the expected diazepine derivative we isolated only the intermediate title compound, (I).



The asymmetric unit of (I) was chosen such that the bromo molecule and the solvent DMF molecule form an $R_2^1(7)$ ring (Bernstein *et al.*, 1997), *via* the N1-H1···O21 and N12-H12···O21 hydrogen bonds (Fig. 1).

In the following discussion, symmetry codes are as in Table 1. The main supramolecular structure is defined by the O12– $H12A \cdots O151^{i}$ hydrogen bond, which links the molecules into a C(11) chain running parallel to the *b* axis. The C21– $H21 \cdots O152^{i}$ interaction forms a link in which the solvent molecule acts a bridge to adjacent molecules in the chain, creating an $R_3^3(12)$ ring.

Centres-of-symmetry at points $(\frac{3}{2}, n + \frac{1}{2}, \frac{1}{2})$, where *n* is any positive or negative integer, produce an antiparallel chain, which is linked to the original by the C13-H13···O12ⁱⁱ and C14-H14···O151ⁱⁱⁱ weak hydrogen bonds. The former, with its symmetry-related bond C13ⁱⁱ-H13ⁱⁱ···O12, form a series of $R_2^2(14)$ rings, centred on centres of symmetry, as described above where *n* is even. The latter, with its symmetry-related

Received 27 January 2003 Accepted 29 January 2003 Online 7 February 2003



Figure 1

A view of (I), with the atomic numbering scheme. Displacement ellipsoids are drawn at the 30% probability level. Hydrogen bonds are indicated by dashed lines.

bond C14ⁱⁱⁱ—H13ⁱⁱⁱ···O12, forms a series of $R_2^2(10)$ rings, centred on centres of symmetry, as described above, where *n* is odd. These two bonds, along with the O12—H12A···O151 hydrogen bond in each antiparallel chain, form a series of staggered $R_3^2(7)$ rings which lie between each of the centro-symmetric rings. This produces molecular ladders, which lie parallel (001) and run along the *b* axis.

Details of the hydrogen bonds are given in Table 1 and a stereoview of this supramolecular structure is shown in Fig. 2.

A similar ladder structure, with molecules in the 'uprights' connected by $N-H\cdots O$ hydrogen bonds and the 'rungs' consisting of weak $C-H\cdots O$ hydrogen bonds, is found in the

structure of 2-trifluoromethyl-4-nitroaniline (Glidewell *et al.*, 2002).

In addition, there is a short contact between methyl atom H22B and $O4^{iv}$.

Examination of the structure with *PLATON* (Spek, 2002) showed that there are no solvent-accessible voids in the crystal lattice.

Experimental

A solution of equimolar amounts of 2-(2-amino-4-nitroanilino)ethanol (700 mg, 3.55 mmol) and 3-dimethylamino-1-(4-bromophenyl)-1-propanone in 15 ml of absolute ethanol was heated to reflux for 30 min. The resulting precipitate was filtered, washed with ethanol and recrystallized from ethanol, yielding red crystals (m.p. 434 K, yield 55%). Analysis calculated for $C_{17}H_{18}BrN_3O_4$: C,50.02, H 4.44, N 10.29%; found: C 50.11, H 4.52, N 10.24%.

Crystal data

 $\begin{array}{l} C_{17}H_{18}BrN_{3}O_{4}\cdot C_{3}H_{7}NO\\ M_{r}=481.35\\ Triclinic, P\overline{1}\\ a=10.5267\ (2)\ \text{\AA}\\ b=10.7688\ (3)\ \text{\AA}\\ c=10.9397\ (3)\ \text{\AA}\\ \alpha=77.646\ (2)^{\circ}\\ \beta=72.743\ (2)^{\circ}\\ \gamma=62.7565\ (12)^{\circ}\\ V=1048.45\ (5)\ \text{\AA}^{3} \end{array}$

Data collection

Nonius KappaCCD diffractometer φ scans and ω scans with κ offsets Absorption correction: multi-scan (*DENZO-SMN*; Otwinowski & Minor, 1997)

 $T_{\min} = 0.599, T_{\max} = 0.670$ 13 949 measured reflections

Refinement

Refinement on F^2 $R[F^2 > 2\sigma(F^2)] = 0.037$ $wR(F^2) = 0.088$ S = 1.054648 reflections 274 parameters H-atom parameters constrained Z = 2 $D_x = 1.525 \text{ Mg m}^{-3}$ Mo K\alpha radiation Cell parameters from 4648 reflections $\theta = 3.0-27.5^{\circ}$ $\mu = 2.00 \text{ mm}^{-1}$ T = 120 (1) KBlock, red $0.30 \times 0.25 \times 0.20 \text{ mm}$

4648 independent reflections 3927 reflections with $I > 2\sigma(I)$ $R_{int} = 0.050$ $\theta_{max} = 27.5^{\circ}$ $h = -13 \rightarrow 13$ $k = -13 \rightarrow 13$ $l = -13 \rightarrow 14$

$$\begin{split} w &= 1/[\sigma^2(F_o^2) + (0.0328P)^2 \\ &+ 0.7369P] \\ \text{where } P &= (F_o^2 + 2F_c^2)/3 \\ (\Delta/\sigma)_{\text{max}} &= 0.001 \\ \Delta\rho_{\text{max}} &= 0.39 \text{ e } \text{\AA}^{-3} \\ \Delta\rho_{\text{min}} &= -0.57 \text{ e } \text{\AA}^{-3} \end{split}$$



Figure 2

A stereoview of the crystal structure of (I). Hydrogen bonds are indicated by dashed lines.

Table 1	
Hydrogen-bonding geometry (Å, °).	

$D - H \cdots A$	D-H	$H \cdots A$	$D \cdots A$	$D - H \cdots A$
N1-H1···O21	0.88	2.17	3.035 (3)	169
$O12-H12E\cdots O151^{i}$	0.84	1.96	2.793 (3)	171
N12-H12···O21	0.88	1.99	2.868 (3)	173
$C13-H13\cdots O12^{ii}$	0.95	2.44	3.332 (3)	155
C14-H14···O151 ⁱⁱⁱ	0.95	2.47	3.315 (3)	148
$C21 - H21 \cdots O152^i$	0.95	2.47	3.407 (3)	171
C22-H22 B ···O4 ^{iv}	0.98	2.50	3.439 (4)	160

Symmetry codes: (i) x, y - 1, z; (ii) 3 - x, -y, 1 - z; (iii) 3 - x, 1 - y, 1 - z; (iv) 1 - x, 1 - y, 2 - z.

H atoms were treated as riding atoms, with C-H = 0.95-0.99 Å, N-H = 0.88 Å and O-H = 0.84 Å. The position of the hydroxy H atom was determined from a difference map and then idealized.

Data collection: *KappaCCD Server Software* (Nonius, 1997); cell refinement: *DENZO–SMN* (Otwinowski & Minor, 1997); data reduction: *DENZO–SMN*; program(s) used to solve structure: *SHELXS*97 (Sheldrick, 1997); program(s) used to refine structure: *SHELXL*97 (Sheldrick, 1997); molecular graphics: *ORTEPII* (Johnson, 1976) and *PLATON* (Spek, 2002); software used to prepare material for publication: *SHELXL*97 and *WordPerfect* macro *PRPKAPPA* (Ferguson, 1999).

X-ray data were collected at the EPSRC X-ray Crystallographic Service, University of Southampton; the authors thank the staff for all their help and advice. JNL thanks NCR Self-Service, Dundee, for grants which have provided computing facilities for this work. MN, AS and JC thank the Ministerio de Educación Cultura y Deportes (Programa de Cooperación con Iberoamérica, AECI) of Spain for financial support for this work. BI and YC thank COLCIENCIAS and Universidad del Valle for financial support.

References

- Bernstein, J., Davis, R. E., Shimoni, L. & Chang, N.-L. (1995). Angew. Chem. Int. Ed. Engl. 34, 1555–1573.
- Ferguson, G. (1999). PRPKAPPA. University of Guelph, Canada.
- Glidewell, C., Low, J. N., McWilliam, S. A., Skakle, J. M. S. & Wardell, J. L., (2002). Acta Cryst. C58, 097–099.
- Johnson, C. K. (1976). *ORTEPII*. Report ORNL-5138. Oak Ridge National Laboratory, Tennessee, USA.
- Nonius (1997). *KappaCCD Server Software*. Windows 3.11 Version. Nonius BV, Delft, The Netherlands.
- Otwinowski, Z. & Minor, W. (1997). *Methods in Enzymology*, Vol. 276, *Macromolecular Crystallography*, Part A, edited by C. W. Carter Jr & R. M. Sweet, pp. 307–326. New York: Academic Press.
- Sheldrick, G. M. (1997). SHELXL97 and SHELXS97. University of Göttingen, Germany.
- Spek, A. L. (2002). *PLATON*. Version of March 2002. University of Utrecht, The Netherlands.